

Write the formulas for the following ionic compounds:

- 1) zinc bicarbonate _____
- 2) cobalt (III) phosphate _____
- 3) gallium selenide _____
- 4) strontium iodide _____
- 5) titanium (IV) fluoride _____
- 6) silver hydroxide _____
- 7) lead (IV) perchlorate _____
- 8) zinc hydride _____
- 9) potassium chromate _____
- 10) iron (III) oxalate _____
- 11) copper (II) bisulfate _____
- 12) vanadium (V) dichromate _____
- 13) lead (IV) thiosulfate _____
- 14) chromium (VI) phosphate _____
- 15) manganese (III) nitrite _____
- 16) tin (IV) carbonate _____
- 17) molybdenum (VI) hydrogen phosphate _____
- 18) calcium hypochlorite _____
- 19) beryllium acetate _____
- 20) platinum (II) arsenide _____

Write the formulas for the following ionic compounds:

- 1) zinc bicarbonate **Zn(HCO₃)₂**
- 2) cobalt (III) phosphate **CoPO₄**
- 3) gallium selenide **Ga₂Se₃**
- 4) strontium iodide **SrI₂**
- 5) titanium (IV) fluoride **TiF₄**
- 6) silver hydroxide **AgOH**
- 7) lead (IV) perchlorate **Pb(ClO₄)₄**
- 8) zinc hydride **ZnH₂**
- 9) potassium chromate **K₂CrO₄**
- 10) iron (III) oxalate **Fe₂(C₂O₄)₃**
- 11) copper (II) bisulfate **Cu(HSO₄)₂**
- 12) vanadium (V) dichromate **V₂(Cr₂O₇)₅**
- 13) lead (IV) thiosulfate **Pb(S₂O₃)₂**
- 14) chromium (VI) phosphate **Cr(PO₄)₂**
- 15) manganese (III) nitrite **Mn(NO₂)₃**
- 16) tin (IV) carbonate **Sn(CO₃)₂**
- 17) molybdenum (VI) hydrogen phosphate **Mo(HPO₄)₃**
- 18) calcium hypochlorite **Ca(ClO)₂**
- 19) beryllium acetate **Be(C₂H₃O₂)₂**
- 20) platinum (II) arsenide **Pt₃As₂**